

PROF. DR J. TINBERGEN

THE HAGUE, 19 January 1954
Sijzenlaan 41

Ho
Miss Khurshid Hasan,
Institute of Social Studies,
Molenstraat 27,
The Hague

Dear Miss Hasan,

My wife and I hope that you are willing to
join Professor Mukerji and Professor and Mrs Lambers to have
dinner with us on January 25.

With best regards,
yours very sincerely,

1708. 1-1. The object of this study is to determine the effect of the various factors on the rate of the reaction.

The object of this study is to determine the effect of the various factors on the rate of the reaction. The factors considered are the concentration of the reactants, the temperature, and the presence of a catalyst.

The rate of the reaction was determined by measuring the amount of product formed in a given time. The concentration of the reactants was varied, and the effect on the rate was observed. The temperature was also varied, and the effect on the rate was observed. The presence of a catalyst was also studied, and its effect on the rate was observed.

The results of the study show that the rate of the reaction is affected by the concentration of the reactants, the temperature, and the presence of a catalyst. The rate increases with increasing concentration of the reactants, with increasing temperature, and with the presence of a catalyst.